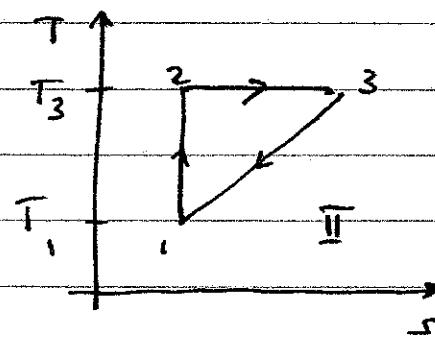
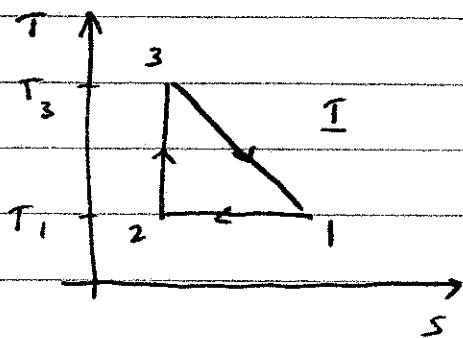


Concepts: def. of entropy, thermal eff.

25



all processes  
internally reversible  
 $Tds = dq$

$$\eta_{th} = \frac{w_{net}}{q_A} = \frac{q_A + q_R}{q_A}$$

$q$  is heat to system ( $q_R < 0$ )  
and  $q = \int T ds$

$$\eta_{th}^I = 1 + \frac{\int_1^2 T ds}{\int_1^3 T ds} = 1 + \frac{\bar{T}_{12}(s_2 - s_1)}{\bar{T}_{31}(s_1 - s_3)} = 1 - \frac{\bar{T}_{12}}{\bar{T}_{31}}$$

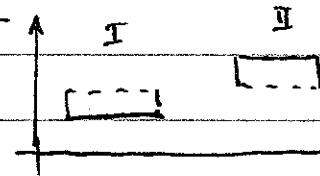
$$\eta_{th}^I = 1 - \frac{2\bar{T}_1}{\bar{T}_3 + \bar{T}_1} \rightarrow \eta_{th}^I = \frac{\bar{T}_3 - \bar{T}_1}{\bar{T}_2 + \bar{T}_1}$$

$$\eta_{th}^II = 1 + \frac{\int_2^3 T ds}{\int_2^1 T ds} = 1 + \frac{\bar{T}_{23}(s_3 - s_2)}{\bar{T}_{23}(s_3 - s_2)} = 1 - \frac{\bar{T}_{31}}{\bar{T}_{23}}$$

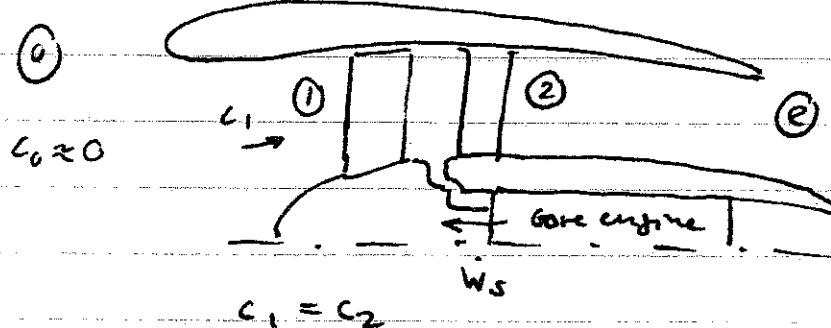
$$\eta_{th}^II = 1 - \frac{\bar{T}_3 + \bar{T}_1}{2\bar{T}_3} \rightarrow \eta_{th}^II = \frac{\bar{T}_3 - \bar{T}_1}{2\bar{T}_3}$$

$\underline{\eta_{th}^I > \eta_{th}^{II}}$  because temperature ratio of heat sink to heat source (on average basis) lower for cycle I

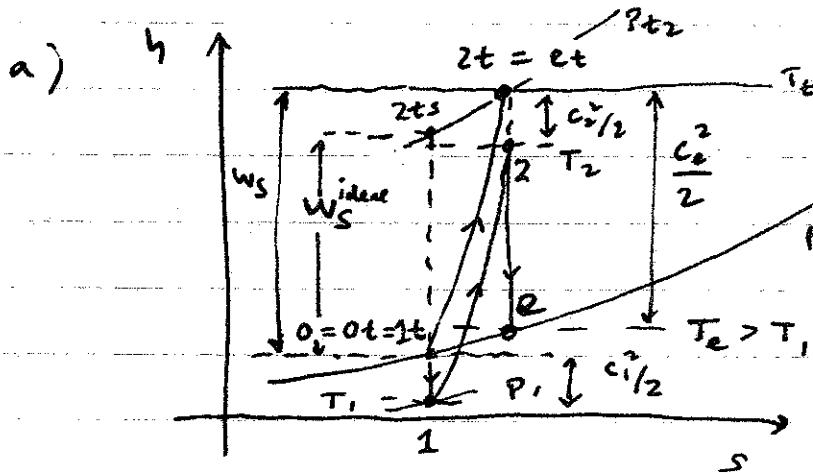
$$\frac{\bar{T}_R}{\bar{T}_A} |_I < \frac{\bar{T}_R}{\bar{T}_A} |_{II}$$



cannot:  
more "waste"  
heat in cycle II  
for same net  
work



$$P_e = P_0$$



$$T_{t2} = T_{ce} \quad \text{5) 1st Law CV enthalpy:}$$

$$W_s = w_s(h_{t2} - h_{t1})$$

$$h_{t1} = h_{t1}, \quad h_{t2} = h_{t2}$$

$$\dot{m} = \frac{w_s}{c_p(T_{t2} - T_{t1})}$$

$$T_{t1} = T_{t2} = T_0 = 300\text{K}$$

use adiab. eff to find  $T_{t2}$ :  $\eta_{fan}^{ad} = \frac{w_s^{ideal}}{w_s} = \frac{T_{t2s} - T_{t1}}{T_{t2} - T_{t1}}, \frac{T_{t2}}{T_{t1}} = \left(\frac{P_{t2}}{P_{t1}}\right)^{\frac{1}{k-1}}$

$$\text{so } T_{t2} = T_{t1} \left(1 + \frac{1}{\eta_{fan}^{ad}} \left[ \left(\frac{P_{t2}}{P_{t1}}\right)^{\frac{1}{k-1}} - 1 \right]\right); \quad T_{t2} = 335.6\text{K}; \quad \dot{m} = 699\text{ kg/s}$$

$$c) \quad T_{t2} = T_2 + \frac{c_2^2}{2c_p}, \quad c_1 = c_2 = \sqrt{2c_p(T_0 - T_1)}, \quad T_2 = T_{t2} - \frac{c_2^2}{2c_p} = 320.5\text{K}$$

$$c_1 = c_2 \rightarrow T_{t1} = T_0 = T_1 + \frac{c_1^2}{2c_p} \quad \uparrow \quad M_2 = \frac{c_2}{\sqrt{RT_2}} \quad M_2 = 0.485$$

$$d) \quad \text{Gibbs: } T_{t2} = dH_t - vdp_t \Big| S; \quad s_2 - s_1 = c_p \ln\left(\frac{T_{t2}}{T_{t1}}\right) - R \ln\left(\frac{P_{t2}}{P_{t1}}\right)$$

no other interaction:  $\Delta S_{tot} = \Delta S_{gen} = \Delta S_{fan} = s_2 - s_1, \quad \alpha_{gen} = 16.1^{\circ}/\text{kg}\cdot\text{K}$   
(ad. fan).

e) Integral mean. & qM:

$$T = \dot{m}(c_e - c_0) = \dot{m}c_e; \quad \downarrow$$

$$T = 173.7 \text{ KN} \quad \downarrow$$

isentropic expansion in nozzle

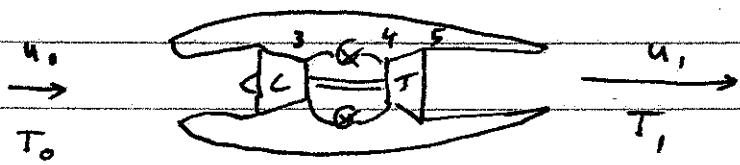
$$P_{t2} = P_{t2} = P_e \left(1 + \frac{k-1}{2} M_e^2\right)^{\frac{1}{k-1}}, \quad P_e = P_{t1}$$

$$M_e = \sqrt{\left[\frac{P_{t2}}{P_{t1}}\right]^{\frac{k-1}{2}}} = 0.71 \quad \left\{ c_e = M_e \sqrt{RT_2} \right.$$

$$T_e = T_{t2} \left(1 + \frac{k-1}{2} M_e^2\right)^{-1} = 304.8\text{K} \quad c_e = 248.5\text{ m/s}$$

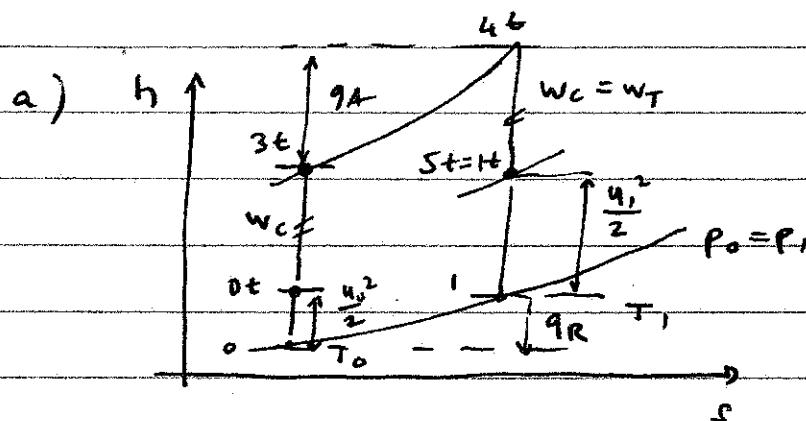
T3

16. Unified Sp 09 25



know:  $u_0, u_1, T_0, T_1, c_p$

assume: ideal cycle  
steady, level flight  
ideal gas,  $c_p = \text{const}$



Concepts: ideal Brayton cycle  
h-s diagrams  
overall, prop. eff.

b)  $q_R = c_p (T_1 - T_0)$  from 1st law

c)  $q_A - q_R = q_{\text{heat}} = w_{\text{net}}$ ,  $w_{\text{net}} = \frac{u_1^2}{2} - \frac{u_0^2}{2}$ ;  $q_{\text{net}} = \frac{u_1^2 - u_0^2}{2}$

d)  $\gamma_{\text{th}} = \frac{w_{\text{net}}}{q_A} = \frac{w_{\text{net}}}{q_R + w_{\text{net}}} = \frac{1}{1 + \frac{q_R}{w_{\text{net}}}}$

$$\gamma_{\text{th}} = \left( 1 + \frac{2c_p(T_1 - T_0)}{u_1^2 - u_0^2} \right)^{-1}$$

e)  $\eta_o = \eta_{\text{prop}} \cdot \gamma_{\text{th}}$ ;  $\eta_{\text{prop}} = \frac{\text{thrust power}}{\text{mass-power}} = \frac{u_1(u_1 - u_0) u_0}{u_1(u_1^2/2 - u_0^2/2)}$

$$\eta_{\text{prop}} = \frac{2}{1 + \frac{u_1}{u_0}}$$

$$\eta_o = \frac{\frac{2}{1 + \frac{u_1}{u_0}} (u_1^2 - u_0^2)}{u_1^2 - u_0^2 + 2c_p(T_1 - T_0)}; \quad \eta_o = \frac{2u_0 \cdot (u_1 - u_0)}{u_1^2 - u_0^2 + 2c_p(T_1 - T_0)}$$